

CHE323/384 Chemical Processes for Micro- and Nanofabrication
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Homework #12

1. An i-line resist has the following properties:

$$A = 0.85 \mu\text{m}^{-1}$$

$$B = 0.05 \mu\text{m}^{-1}$$

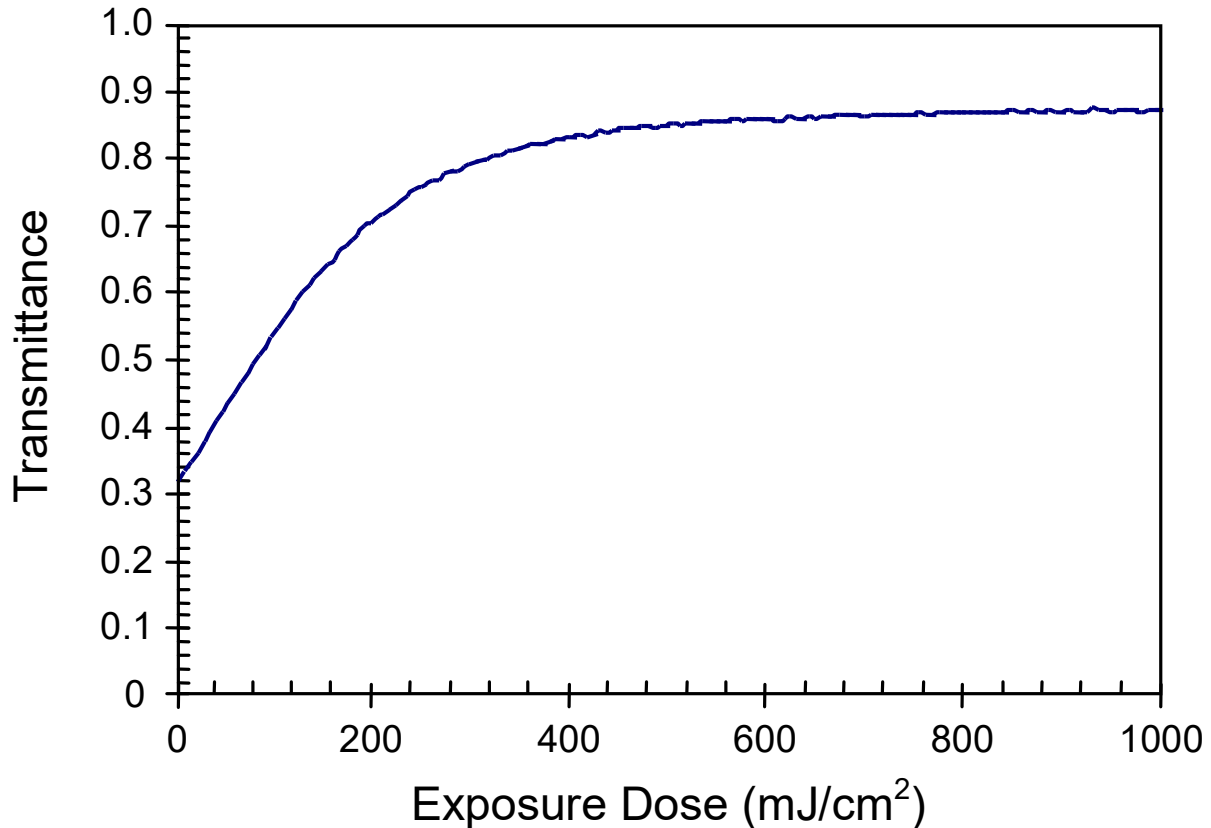
$$C = 0.018 \text{ cm}^2/\text{mJ}$$

$$\text{Refractive index} = 1.72$$

The resist is coated to a thickness of $1.1 \mu\text{m}$ on a glass substrate optically matched to the photoresist. At the beginning of exposure, what percentage of the incident light makes it to the bottom of the resist?

Note that $T_{12} = 1 - \left(\frac{n_2 - n_1}{n_2 + n_1}\right)^2$

2. From the transmittance curve below, estimate the values of A , B and C . The resist thickness used was $0.75 \mu\text{m}$ and the measurement was performed in the standard way. Assume a typical i-line resist with refractive index = 1.69.



3. For a chemically amplified resist (and ignoring the effects of diffusion and acid loss on concentration),

$$h = 1 - e^{-CI t}$$

$$m = e^{-K_{amp} t_{PEB} h}$$

From these equations,

- Derive an expression for the relative bake time sensitivity of m (i.e., calculate $dm/d\ln t_{PEB}$).
- Derive an expression for the relative temperature sensitivity of m (i.e., calculate $dm/d\ln T$). From this, will a low activation energy resist or a high activation energy resist be more sensitive to temperature variations?
- Does the presence of base quencher change the bake time or temperature sensitivity of m ?

4. Why does the addition of base quencher reduce the sensitivity of the resist to airborne base contaminants?