

CHE323/CHE384  
 Chemical Processes for Micro- and Nanofabrication  
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## Lecture 13 Diffusion, part 1

Chris A. Mack  
 Adjunct Associate Professor

**Reading:**  
 Chapter 3, *Fabrication Engineering at the Micro- and Nanoscale*, 4<sup>th</sup> edition, Campbell

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## Diffusion – Historical Perspective

- The term ‘diffusion’ has been used in two different ways in the semiconductor industry
  - Dopant introduction
  - Dopant redistribution
- Today, most processes introduce dopants into the wafer via ion implantation
  - Diffusion refers to the redistribution of dopants after implant

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## Diffusion as Dopant Introduction

- Performed in a furnace (similar to oxidation)
- Dopant introduced to wafers in three ways:
  - Solid source: disks of BN, AlAsO<sub>4</sub>, etc., placed between wafers in the carrier
  - Liquid source: POCl<sub>3</sub>, BBr<sub>3</sub>, etc., with N<sub>2</sub> bubbler (most common)
  - Gas Source

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## Diffusion as Dopant Introduction

- Dopant introduced at top of wafer, then diffuses in
- Run at concentrations exceeding the solid solubility limit in silicon to improve control
- Applications:
  - Device generations older than 2 μm
  - Bipolar high dopant steps (emitter)
  - Polysilicon doping

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## Diffusion as Dopant Redistribution

- Initial distribution of dopant supplied, usually by ion implantation
  - Ion implantation destroys the crystal structure, creating regions of amorphous silicon
  - High temperatures are required to regrow the crystal and activate the dopant (annealing)
  - These high temperatures result in diffusion of the dopants
- Our goal: predict the final dopant distribution

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## Diffusion Review

- Fick's First Law (in one dimension):
 
$$\text{material flux } \rightarrow J = -D \frac{\partial C(x, t)}{\partial x}$$

↖ Dopant concentration  
↘ Diffusivity of dopant in silicon
- Continuity equation (mass balance)
 
$$\frac{\partial C(x, t)}{\partial t} = -\frac{\partial J}{\partial x}$$

Change in flux causes accumulation of material

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## Diffusion Review

- Combining Fick's first law with the continuity equation gives Fick's 2<sup>nd</sup> law (in 1-D):
 
$$\frac{\partial C(x, t)}{\partial t} = \frac{\partial}{\partial x} \left( D \frac{\partial C(x, t)}{\partial x} \right)$$
- In 3-D:  $\frac{\partial C}{\partial t} = \nabla(D\nabla C)$ ,  $\nabla = \frac{\partial}{\partial x} + \frac{\partial}{\partial y} + \frac{\partial}{\partial z}$

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## Solving the Diffusion Equation

- To solve the diffusion equation, we need
  - One initial condition
  - Two boundary conditions for each dimension
  - Knowledge of the diffusivity  $D(x,y,z,t)$  or  $D(C)$
- The real complications come from knowing the diffusivity
  - If the diffusivity is constant, solutions are easy

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## Lecture 13: What have we learned?

- What are the two meanings of the term 'diffusion' in semiconductor processing?
- Explain how dopants are introduced during an old-style diffusion step
- Why is dopant diffusion inevitable after ion implantation?
- What does one need to know in order to solve the diffusion equation?

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